

Thermodynamics of nano-cluster phases: a unifying theory

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We propose a unifying, analytical theory accounting for the self-organization of colloidal systems in nano- or micro-cluster phases. We predict the distribution of cluster sizes with respect to interaction parameters and colloid concentration. In particular, we anticipate a proportionality regime where the mean cluster size grows proportionally to the concentration, as observed in several experiments. We emphasize the interest of a predictive theory in soft matter, nano-technologies and biophysics.

There is an increasing interest in colloid science and emerging nano-technologies for systems exhibiting self-assembled, nano-structured spatial patterns at equilibrium (pseudo-periodic structures in 2D or 3D, such as stripes, ripples, bubbles, lamellae, tubes or clusters in a large variety of systems) [1, 2]. A competition between a short-range attractive interaction (*e.g.* a depletion force) which favors condensation, and a weaker, longer-range repulsion (*e.g.* electrostatic) which prevents a complete phase separation, leads to equilibrium structures with a characteristic length-scale (the pattern typical size). Here we focus on *cluster phases*, a particular type of space patterns, consisting of aggregates of sizes ranging from a few to hundreds of particles. They are the colloid analog of micelles [3] and occur for a variety of physical systems (*e.g.* colloids, star polymers, proteins) and interactions [1, 4, 5, 6, 7, 8, 9, 10, 11, 12], whenever the strength of the interactions is of the order of the thermal energy $k_B T$. Such patterns have also been experimentally observed or simulated in two dimensions [2, 10, 12, 13]. Recently, it has been proposed by one of us that proteins embedded in cell membranes can also be found in two-dimensional cluster phases [14]. Indeed, they experience repulsive and attractive forces mediated by the lipidic membrane, the range of which is nanometric [15, 16, 17]. Nowadays, such cluster structures can be observed in cell plasma membranes with the help of advanced microscopy techniques [18, 19].

It is of great interest in this context to be able to predict *a priori* for which regimes of parameters one can expect a cluster phase to exist, and to anticipate the mean cluster size. However, apart from generic phenomenological arguments [2, 5], the thermodynamics of cluster phases has only been tackled in specific frameworks and dimensions, ranging from Van der Waals approximative schemes [9] to electrostatic models taking explicitly into account counter-ions [6, 20], or estimations of structure factors [11, 21]. By contrast, we propose here a general approach based on simple thermodynamical principles, requiring no specific description of microscopic interactions, but simply the existence of some generic features of these interactions, and applicable in two and three di-

mensions. The analytical methods appeal to ideas from elementary micellisation or nucleation theories [3, 22]. They are also reminiscent of Ref. [23], even though this study was limited to steric repulsion between membrane proteins (see also [18]). Up to basic prerequisites, we demonstrate that the cluster phase exists above a critical particle volume fraction, ϕ^c (Fig. 1). Then a gas phase coexists with large clusters. In addition, in several circumstances, the mean aggregation number $\langle k \rangle$ has been measured experimentally in function of the volume fraction ϕ ; $\langle k \rangle$ is extracted either from direct enumeration in confocal microscopy [1, 7] or from structure factors in diffraction experiments [1, 4]. It is found that for a wide regime of concentrations, $\langle k \rangle$ grows proportionally to ϕ :

$$\langle k \rangle \simeq \phi / \phi^c. \quad (1)$$

Our analytical treatment provides a straightforward explanation for this *proportionality regime* (Fig. 1).

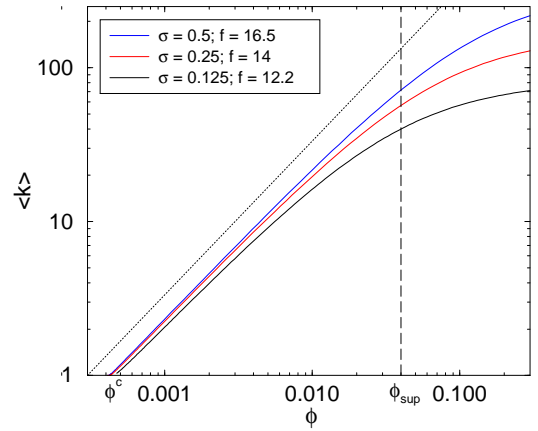


FIG. 1: Exact mean aggregation number $\langle k \rangle$ vs colloid volume fraction ϕ for different sets of parameters, in log-log coordinates, calculated within our framework (Eqs. (6,7)). Here $d = 2$, $\alpha = 3/2$, $\gamma = 40$ and f and σ are indicated in the legend (see definitions in the text and Eq. (14)). The dotted line has slope 1, for comparison. For $\phi < \phi^c$, the systems is in the gas phase and $\langle k \rangle \simeq 1$. The proportionality regime where $\langle k \rangle \propto \phi$ appears clearly for $\phi^c < \phi \lesssim \phi_{\text{sup}}$, followed by a saturation for $\phi > \phi_{\text{sup}}$ where large multimers dominate.

The cluster size distribution is investigated in a statistical mechanics formalism. A useful introduction to this issue can be found in the Appendix of Ref. [22]. We start from the canonical partition function for a system of N interacting particles in a volume V of dimensionality d ,

$$Z = \frac{\Lambda^{-dN}}{N!} \int_{V^N} \mathbf{dr}_1 \dots \mathbf{dr}_N e^{-U_N}, \quad (2)$$

where U_N is total interaction energy of the N particles; here and in the sequel, all energies are in units of $k_B T$. Usually, Λ is chosen equal to the de Broglie thermal wavelength $\Lambda = \sqrt{2\pi\hbar^2/mk_B T}$. However, for sake of convenience, we shall choose it to be the particle diameter. Suppose now that one is able to determine unambiguously regions V_k of V that partition the N particles into clusters: N_1 monomers, N_2 dimers, etc. . . , each k -mer in a distinct region V_k , so that $N = \sum k N_k$. Then the integral (2) can be written as a sum of integrals on the V_k , because interactions between the regions are negligible. Simple algebra [22] leads to

$$Z = \sum_{\{N_k\}} \prod_k \frac{1}{N_k!} \left(V \Lambda^{-d} e^{-F(k)} \right)^{N_k}, \quad (3)$$

where $F(1) = 0$ and

$$F(k) = -\ln \left\{ \frac{\Lambda^{d(1-k)}}{k!} \int_{V_k} \mathbf{dr}_1 \dots \mathbf{dr}_{k-1} e^{-U_k} \right\} \quad (4)$$

for $k > 1$. It is the free energy of a k -cluster.

The particle organization is described by the mean volume fraction of k -clusters which derives from Z ,

$$c_k \equiv \frac{\langle N_k \rangle \Lambda^d}{V} = e^{\mu k - F(k)} = c_1 e^{-G(k, \mu)}, \quad (5)$$

where we have introduced the chemical potential $\mu = \ln c_1$ and the grand potential of a k -cluster $G(k, \mu) = F(k) - (k-1)\mu$. The value of μ , or equivalently of the monomer fraction c_1 , is fixed by the constraint [3, 22]

$$\phi \equiv \frac{N \Lambda^d}{V} = \sum_{k=1}^{\infty} k c_k. \quad (6)$$

We also define the total volume fraction of clusters (including monomers) M , the total fraction of multimers \hat{M} and the mean cluster aggregation number $\langle k \rangle$,

$$M \equiv \sum_{k=1}^{\infty} c_k, \quad \hat{M} \equiv \sum_{k=2}^{\infty} c_k \quad \text{and} \quad \langle k \rangle = \frac{\phi}{M}. \quad (7)$$

The interaction energy of the k particles of a cluster, U_k , is the sum of two contributions: U_0 due to the short range interactions (hard core repulsion at contact and short range attraction) and U_r due to the longer range repulsions. The free energy can be split in two parts,

$F(k) = F_0(k) + F_r(k)$; F_0 is the free energy given by (4) in the absence of repulsion ($U = U_0$) and $F_r(k) = -\ln \langle e^{-U_r} \rangle_0$, with $\langle \dots \rangle_0 = e^{F_0(k)} \int \mathbf{dr}_1 \dots \mathbf{dr}_{k-1} (\dots) e^{-U_0}$, is the specific contribution arising from the repulsion.

If particle repulsion is switched-off, the cluster free energy reduces to that of a spherical droplet of simple liquid. It is usually written within a good approximation as the sum of a bulk and surface energy, for $k \gg 1$:

$$F_0(k) = -f_0(k-1) + \gamma(k-1)^{\frac{d-1}{d}}. \quad (8)$$

The free energy per particle $-f_0 < 0$ accounts for the mean energy $e_0 \approx -\varepsilon_a \nu / 2$, where ν is the typical number of neighbors of a particle in the cluster and ε_a the strength of the short range attraction potential between two particles, and an entropic contribution $-k_B [\ln(v_f \Lambda^{-d}) + 1]$ (v_f is the free volume per particle accessible to the particles inside the clusters). For intermediate, finite, values of k , a positive surface correction must be added to the previous bulk contribution. It takes into account the fact that the surface particles have typically twice fewer neighbors than the bulk ones and a larger free volume.

Now we consider that a weak repulsive pair potential is also acting on the particles,

$$U_r(k) = \frac{1}{2} \sum_{i \neq j} v(\mathbf{r}_i - \mathbf{r}_j). \quad (9)$$

If we assume that those weak interactions do not modify the cluster structure and that the particle density, $\rho = k \Lambda^d / V_k \approx 1$, is homogeneous in the cluster, we can make the two successive mean-field approximations:

$$F_r(k) \simeq \langle U_r \rangle_0 \simeq (k-1) \frac{\rho}{2V_k} \int_{V_k \times V_k} \mathbf{dr}_1 \mathbf{dr}_2 v(\mathbf{r}_1 - \mathbf{r}_2). \quad (10)$$

As an example, we consider the following potential of intermediate range $\lambda > \Lambda$,

$$v(r) = \varepsilon_r \left(\frac{\lambda}{r} \right)^{(2-\alpha)d} e^{-r/\lambda}. \quad (11)$$

For $r < \lambda$ it behaves like a long range potential if $\alpha > 1$. But at larger distances, the potential rapidly vanishes. For particles undergoing screened Coulomb repulsion, λ is the Debye screening length, $\alpha = 5/3$ in 3D and, $\alpha = 3/2$ if the particles are confined in 2D, like charged proteins in a membrane. Using the potential (11), by estimating the integral in the repulsive part of the free energy (10), one gets $F_r(1) = 0$ and,

$$F_r(k) \simeq f_r(k-1) + \sigma(k-1)^\alpha, \quad k \ll \rho \lambda^d / \Lambda^d, \quad (12)$$

$$F_r(k) \simeq f_r(k-1), \quad k \gg \rho \lambda^d / \Lambda^d, \quad (13)$$

with $\sigma \sim \varepsilon_r (\rho^{1/d} \lambda / \Lambda)^{(2-\alpha)d}$.

First we consider that the repulsion is of infinite range ($\lambda \rightarrow \infty$) so that (12) holds for any k . According to Eqs. (8,12) the grand potential reads

$$G(k, \mu) = -(f + \mu)(k-1) + \gamma(k-1)^{\frac{d-1}{d}} + \sigma(k-1)^\alpha, \quad (14)$$

with the bulk free energy $f = f_0 - f_r$. The existence of clusters even at significantly low concentrations ($\phi^c \ll 1$) requires $f \gg 1$ to overcome translational entropy (see proof below and Fig 1). We also assume that the repulsion is weak: $\epsilon_r \ll 1$ [14], so that $\sigma < 1$ and $f \gg \sigma$. Finally, we shall see below that the existence of the proportionality regime requires $\gamma \gg \sigma$. The global shape of $G(k)$ – and thus of c_k , Eq. (5) – is very sensitive to the value of μ , which is itself fixed by the total concentration of particles (Eq. 6)); Eqs. (5,6) show that $\mu(\phi)$ increases monotonously with ϕ . The particle concentration thus controls directly the distribution shape.

At low particle concentration ϕ , $\mu = \ln c_1 < \ln \phi$ takes a large negative value and then, $G(k)$ increases monotonously with k . It follows from (5) that the cluster size distribution c_k is maximal at $k = 1$ and decreases exponentially with k . Since the behavior of $G(k)$ is dominated by $-\mu k$, the width of the distribution c_k goes as μ^{-1} . The volume contains mainly single particles and very few transient and small clusters formed by thermal fluctuations. The particles form a *gas phase*. The mean cluster size, $\langle k \rangle \simeq 1$, is slightly above 1.

Above a critical concentration ($\phi > \phi^c$), the chemical potential $\mu > \mu^c$ is such that $G(k)$ has a local minimum at $k^* > 1$. In this regime, the distribution c_k is bimodal with two maxima at $k = 1$ and $k = k^*$ [14]. The particles are partitioned between a gas of monomers and stable clusters of aggregation number fluctuating about k^* . Such a configuration is often called a *cluster phase*. Since the distribution is peaked at k^* , we can estimate with a good accuracy the multimer concentration

$$\hat{M} \simeq c_1 e^{-G(k^*, \mu)}, \quad (15)$$

and we write $M = c_1 + \hat{M}$, $\phi \simeq c_1 + k^* \hat{M}$.

At the critical point $\mu = \mu^c$, $G(\mu^c, k)$ has an inflexion point at $k = k^* = k^c$. It follows that μ^c and k^c satisfy $\partial_k G(k^c, \mu^c) = \partial_k^2 G(k^c, \mu^c) = 0$ and then read,

$$k^c - 1 = \left(\frac{1}{\alpha(\alpha-1)} \frac{d-1}{d^2} \frac{\gamma}{\sigma} \right)^{\frac{1}{\alpha-1+1/d}}, \quad (16)$$

$$\mu^c = -f + A\gamma^{\frac{d(\alpha-1)}{1+d(\alpha-1)}} \sigma^{\frac{1}{1+d(\alpha-1)}}, \quad (17)$$

where $A > 0$ is a long prefactor, function of α and d . The condition $\gamma \gg \sigma$ ensures that $k^c \gg 1$. The requirement $\phi^c \ll 1$ implies $\mu^c = \ln c_1^c < \ln \phi^c \ll -1$. Eq. (17) implies $f > -\mu^c$, which justifies the condition $f \gg 1$ above. In addition, $G(k^c, \mu^c) \gg 1$ thus $\hat{M}^c \ll c_1^c$ and

$$\phi^c \simeq c_1^c \simeq M^c. \quad (18)$$

In the cluster phase close to the critical point, $k^* \simeq k^c$ and, since $G(k^c, \mu^c) \gg 1$, the concentration of monomers exceeds by far that of multimers, $\hat{M} \ll c_1$ (15). As we go deeper into the cluster phase by increasing ϕ or μ , the typical cluster size k^* shifts to larger values and \hat{M} raises as the energy well $G(k^*, \mu)$ deepens. In order to get quantitative insights on the behaviors of the cluster phase above the critical point, among which the demonstration of the proportionality law (1), we perform a systematic expansion in terms of the small parameter $(k^* - k^c)/k^c > 0$. Since those calculations involve very long prefactors in the general case, we treat here the particular case $d = 2$ and $\alpha = 3/2$ [2, 14, 23]; for other values of d and α , the calculation can easily be performed following the same route. In this case, we have $k^c = 1 + \gamma/3\sigma \simeq \gamma/3\sigma \gg 1$, $\mu^c = -f + \sqrt{3\sigma\gamma}$ and $G(k^c, \mu^c) = \frac{1}{3\sqrt{3}} \frac{\gamma^{3/2}}{\sigma^{1/2}} \gg 1$. First the deviation of the chemical potential from its critical value is obtained by doing the expansion of the equation satisfied by k^* at a given μ , $\partial_k G(k^*, \mu) = 0$, in terms of $\mu - \mu^c$ and $k^* - k^c$ up to the first significant order. Thus

$$\mu - \mu^c \simeq \frac{\sqrt{3}}{8} \sqrt{\gamma\sigma} \left(\frac{k^* - k^c}{k^c} \right)^2. \quad (19)$$

Next, using (19), we can obtain the increase of the cluster free energy, $\Delta G = G(k^*, \mu) - G(k^c, \mu^c)$, as

$$\Delta G \simeq -\frac{3}{8} G(k^c, \mu^c) \left(\frac{k^* - k^c}{k^c} \right)^2. \quad (20)$$

It allows us to express the increase of the monomer and multimer concentrations from their values at the critical point $c_1^c = e^{\mu^c}$ and $\hat{M}^c = c_1^c e^{-G(k^c, \mu^c)}$

$$c_1 - c_1^c \simeq c_1^c (\mu - \mu^c), \quad (21)$$

$$\hat{M} - \hat{M}^c \simeq \hat{M}^c (\mu - \mu^c + e^{-\Delta G} - 1). \quad (22)$$

According to Eqs. (19-22), we finally obtain the increase of the total cluster concentration $M = c_1 + \hat{M}$:

$$M - M^c = c_1 - c_1^c + \hat{M} - \hat{M}^c, \quad (23)$$

$$\simeq M^c (\mu - \mu^c) + \hat{M}^c (e^{-\Delta G} - 1), \quad (24)$$

as well as the increase of $\phi \simeq c_1 + k^* \hat{M}$:

$$\begin{aligned} \phi - \phi^c &\simeq c_1 - c_1^c + k^c (\hat{M} - \hat{M}^c) + \hat{M}^c (k^* - k^c) \\ &\simeq \phi^c (\mu - \mu^c) + \\ &\quad + k^c \hat{M}^c \left(e^{-\Delta G} - 1 + \frac{k^* - k^c}{k^c} \right). \end{aligned} \quad (25)$$

We have expanded G in powers of $k - k^c$ at the lowest significant order. One can prove that these expansions are relevant while $k - k^c < k^c$. In contrast, expansions of e^{-G} become rapidly erroneous when $k - k^c = \mathcal{O}(k^c)$. Thus we keep the exponentials in the expansions.

The inspection of Eqs. (20,24,26) reveals that when $k^* - k^c$ grows, the total fraction of colloids, ϕ , can increase considerably without the total fraction of clusters,

M , varying significantly. It follows that $\langle k \rangle = \phi/M$ grows linearly with ϕ within a very good approximation, which proves the proportionality law (1). More precisely, $M - M^c$ remains of the order of M^c provided that (i) $\mu - \mu^c < 1$, i.e. $k^* - k^c < (8/\sqrt{3\gamma\sigma})^{1/2} k^c$; and (ii) $\hat{M}^c e^{-\Delta G} < M^c$, i.e. $G(k^c, \mu^c) + \Delta G > 0$ or $k^* - k^c < 2\sqrt{2}/3k^c$. We denote by k_{sup} the ensuing limiting value: $k_{\text{sup}} - k^c = \inf \left[\left(\frac{8}{\sqrt{3\gamma\sigma}} \right)^{1/2}, \frac{2\sqrt{2}}{3} \right] k^c$. For reasonable parameter values $\sigma < 0.5$ and $\gamma < 100$, such as in Fig. 1, the numerical prefactor above is close to 1 or larger. We simplify below this condition to $k_{\text{sup}} = 2k^c$.

By contrast to M , ϕ grows rapidly with μ because of the prefactor k^c in the second term of Eq. (26). When $k = k_{\text{sup}}$, one gets $\phi_{\text{sup}} \simeq k^c \phi^c$ by Eq. (26). Thus $\phi_{\text{sup}}/\phi^c = k_{\text{sup}}/2 \simeq \langle k \rangle$, because at this ϕ , about one half of the clusters are monomers and the other half are multimers of size k_{sup} . Therefore the proportionality regime remains valid up to $\phi = \phi_{\text{sup}}$, as illustrated in Fig. 1.

To sum up, there are 3 regimes: if $\phi < \phi^c \simeq c_1^c = e^{\mu^c}$, the system essentially contains monomers; if $\phi > \phi_{\text{sup}}$, far deep in the cluster phase, where the fraction of clusters exceeds the concentration of single particle, $\hat{M} > c_1$, the average aggregation number is obviously $\langle k \rangle \simeq k^*$. It varies hardly with ϕ . Finally, the proportionality regime is the range of concentrations $\phi^c < \phi < \phi_{\text{sup}}$ where $\langle k \rangle \simeq \phi/\phi^c \simeq e^{-\mu^c} \phi$, in other words, owing to Eq. (7), where the cluster volume fraction M varies very slowly. The proportionality regime remains valid while the distribution c_k is truly bimodal, that is to say the fraction of multimers remains comparable to the fraction of monomers [24]. This regime is visible in Fig. 1. An estimate of ϕ_{sup} is $\phi_{\text{sup}} \simeq k^c \phi^c = \frac{\gamma}{3\sigma} \phi^c \gg \phi^c$. The proportionality indeed covers about two decades in Fig. 1.

It is also worth discussing the role of small multimers that we have neglected so far. In our framework, monomers are far more numerous than dimers (or small multimers): $c_2/c_1 = \exp(\mu^c - \mu) \exp(\gamma - f + \sigma - \mu) = \exp(\mu^c - \mu) \exp(\gamma + \sigma - \sqrt{3\sigma\gamma}) > \exp(\mu^c - \mu) \exp(\gamma/4)$. Thus $c_2/c_1 \gg 1$ if $\mu \simeq \mu^c$, because $\gamma \gg 1$. This important point justifies the use of the grand potential (14) for small clusters. Indeed, the form of $F(k)$ was *a priori* valid for large k only, where the definition of a surface tension γ and of a “bulk” energy f_0 is meaningful. A correct modeling of dimers would involve the true binding free energy, $F_b(2)$, of a single bond between monomers. With $F_b(2)$ of a few $k_B T$, one also gets that $c_2/c_1 = e^{[F_b(2)] + \mu} \ll 1$, because $\mu \ll -1$. The same arguments hold for c_k/c_1 when $1 < k \ll k^c$. Small multimers are negligible and their exact modeling is irrelevant in our formalism.

Finally, we now explore the case of physical interest λ finite. Eq. (13) shows that at large k , F recovers a simple spherical droplet form, as in Eq. (8). If $f = f_0 - f_r > 0$, then the equilibrium configuration at the large N limit is a single large cluster coexisting with gas, because this configuration saves the surface free-energy cost (if

$f < 0$, the condensation is never favorable). Inspecting Eqs. (12-14), if there exist $\tilde{k} \sim \lambda^d/\Lambda^d$, $\tilde{k} > k^*$ where G is maximum and such that $G(\tilde{k}, \mu) - G(k^*, \mu) \gg k_B T$, the cluster phase becomes metastable.

Thus we have shown that the different physical quantities of interest can be predicted by our analytical approach. Starting from the microscopic interaction potential, we predict a range of concentration where (i) a gas of monomers coexists with large clusters, and (ii) the mean aggregation number is proportional to the colloid concentration, as observed in experiments. We believe that, playing with the effective interaction parameters, in particular the colloid electrostatic repulsion, the Debye screening length and the attractive part due to depletion forces, our predictions can be tested experimentally by confocal microscopy where cluster statistics can be obtained with a very good accuracy [1].

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- [1] A. Stradner, A., *et al.*, Nature **432**, 492 (2004).
 - [2] M. Seul and D. Andelman, Science **267**, 476 (1995).
 - [3] S.A. Safran, Statistical thermodynamics of surfaces, interfaces, and membranes (Perseus, Cambridge, 1994).
 - [4] P.N. Segrè, *et al.*, Phys. Rev. Lett. **86**, 6042 (2001).
 - [5] F. Sciortino, *et al.*, Phys. Rev. Lett. **93**, 055701 (2004).
 - [6] J. Groenewold and W.K. Kegel, J. Phys. Chem. B **105**, 11702 (2001).
 - [7] H. Sedgwick, S.U. Egelhaaf, and W.C.K. Poon, J. Phys.: Condens. Matter **16**, S4913 (2004).
 - [8] E. Stiakakis, *et al.*, Europhys. Lett. **72**, 664 (2005).
 - [9] R.P. Sear and W.M. Gelbart, J. Chem. Phys. **110**, 4582 (1999).
 - [10] R.P. Sear, *et al.*, Phys. Rev. E **59**, R6255 (1999).
 - [11] Y. Liu, W.R. Chen, and S.H. Chen, J. Chem. Phys. **122**, 044507 (2005).
 - [12] A. Imperio and L. Reatto, J. Phys.: Condens. Matter **16**, S3769 (2004).
 - [13] T. Gulik-Krzywicki, M. Seigneuret, and J.L. Rigaud, J. Biol. Chem. **262**, 15580 (1987).
 - [14] N. Destainville, Phys. Rev. E **77**, 011905 (2008).
 - [15] O.G. Mouritsen, Life - as a matter of fat (Springer, Berlin, Germany, 2005).
 - [16] A. Borodich, I. Rojdestvenski, and M. Cottam, Biophys. J. **85**, 774 (2003).
 - [17] M. Goulian, R. Bruinsma, and P. Pincus, Europhys. Lett. **22**, 145 (1993); J.B. Fournier and P.G. Dommersnes, Europhys. Lett. **39**, 681 (1997).
 - [18] J.J. Sieber, *et al.*, Science **317**, 1072 (2007).
 - [19] P.S.H. Park and K. Palczewski, Nature Chem. Biol. **1**, 184 (2005).
 - [20] J. Groenewold and W.K. Kegel, J. Phys.: Condens. Matter **16**, S4877 (2004).
 - [21] M. Tarzia, and A. Coniglio, Phys. Rev. Lett. **96**, 075702 (2006).
 - [22] D.J. Mitchell and B.W. Ninham, J. Chem. Soc. Trans. 2 **77**, 601 (1981).

[23] P. Sens and M.S. Turner, Biophys J. **86**, 2049 (2004).

[24] Of course, this description fails above a critical concentration $\phi_g \sim 1$ where the system is closely packed or

experiences colloidal gelation [4, 5, 21].